Vithalbhai Patel & Rajratna P. T. Patel Science College Vallabh Vidyanagar B. Sc. (Sixth Semester) (CHEMISTRY) Subject : ORGANIC CHEMISTRY (US06CCHE01)Date : 05-03-2019Internal Examination - 2019Day : TuesdayTime : 10.00 to 12.00 noon Note: (i) All questions are to be attempted. (ii) Figures to the right indicate marks.		
Q.1	Choose the correct option for the following : [8]	
(i)	(-) - Arabinose is converted into D – (+) – glucose by	
	(a) Killiani-Fisher synthesis (b) Ruff degradation	
	(c) Haworth synthesis (d) none of these	
(ii)	Which one is an invert sugar ?	
	(a) (+) - Glucose (b) (+) - Maltose (c) (+) - Mannose (d) (+) - Sucrose	
(iii)	Anthracene upon treatment with K ₂ Cr ₂ O ₇ / H ₂ O gives	
	(a) Benzoic acid (b) Phthalic acid (c) Anthraquinone (d) Benzophenone	
(iv)	Molecular formula of naphthalene is	
	(a) C_8H_{10} (b) $C_{10}H_8$ (c) $C_{10}H_{10}$ (d) $C_{10}H_{12}$	
(v)	Which of the following molecule is aromatic in nature ?	
	(a) cyclopentadienyl anion (b) cyclopentadienyl cation	
	(c) cyclopentadienyl radical (d) none of these	
(vi)	In Diels-Alder reaction, the diene must be in configuration.	
	(a) E (b) s-trans (c) s-cis (d) Z	
(vii)	Which group is an auxochrome?	
	(a) $-N=N-(azo)$ (b) $-NH_2$ (c) $-NO_2$ (d) all of these	
(viii)	Which one is used as food dye?	
*	(a) picric acid (b) alizarin (c) methylene blue (d) Orange-I	
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Q.2 (i)	Explain : Phenylhydrazine could be used as a powerful tool in the study of carbohydrates.	
(ii) (iii) (iv) (v) (vi)	Explain that (+)-glucose is a pyranose and not a furanose. Explain : C_1 - C_2 bond is shorter than C_2 - C_3 bond in naphthalene. Write a note on Bucherer reaction. Write Woodward-Hoffmann rules for electrocyclic reaction. Draw molecular orbital diagram of 1,3-butadiene.	

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(vii)	Define Dyes. What are the requisites for a true dye?
(viii	Explain : Crystal violet is lighter in color than malachite green.
Q.3	[08]
[A] [B]	Prove that (+) - glucose is an aldohexose sugar. Show that (+)-maltose is a 4-O-(α-D-glucopyronosyl)-D- glucopyral.cse. OR
Q.3 [A] [B]	[08] Discuss the host-guest relationship properties of cyclodextrin. What are the effect of cyclodextrin on chemical reaction ? Discuss Ruff degradation with suitable illustration.
Q.4	[08]
[A]	Write the structure and name of potent carcinogens. Discuss the mechanism
[B]	Explain: 1-nitronaphthalene upon further nitration gives 1, 8-dinitro naphthalene as a major product.
Q.4	OR [08]
[A] [B]	Give the synthesis of 1, 4, 9 - trimethyl phenanthrene from 1-methyl naphthalene and succinic anhydride using Hawarth synthesis. Discuss Sulfonation of naphthalene.
Q.5 [A]	[08] Explain the suprafacial and antrafacial modes of cycloaddition reaction. Also
	give Woodward-Hoffmann rules for the same.
[B]	Complete and rewrite the following reaction :
	(i) cis, trans-2,4-hexadiene thermal→
	(ii) trans, trans-2,4-hexadiene + ethylene thermal \therefore
0.5	OR [08]
[A] [B]	Explain : [4+2] thermal cyclization takes place readily but [2+2] thermal cylization dose not. Write a note on Cope rearrangement and Sigmatropic reaction.
Q.6	Write synthesis and application of following : (i) Crystal violet (ii) Disperse Blue-1 (iii) Mercurochrome. [08] OR
Q.6 [A]	[08] Give the difference between dyes and pigments. What are the requirements
[B]	of an organic-pigments ? Give synthesis of : (i) Dye used as thermally stable pigment and (ii) Dye used in carbonet beverages

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