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SEAT NO.



No. of printed pages. 03

B.Sc.EXAMINATION -2022 (VIth SEMESTER)

B.Sc.Sem -6 US06CCHE24 (Applied Chemistry)

Date: 7-04-2022, Thursday

Time: 03:00 P.M. To 05:00 P.M.

Total Mark-70

Note: (i) All questions are to be attempted.(ii) Figures to the right indicate marks.

Q=1. Choose the correct option for the following.

(10)

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1. How many CMR signals would you expect from m-Xylene?

(a) 5 (b) 7 (c) 6 (d) 9

2. Which of the following compound have smallest proton chemical shift ?

(a) CH₃Cl (b) R_2 CHCl (c) R_3 C-H (d) RCH₃

3. Which of the following is used as reference in the NMR spectra?

(a) Tetramethylsilane (b) Tetrahydrofuran (c) Dimethylformamide (d) Benzenehexachloride

4. ----- is used in the treatment of tuberculosis.

(a) Acedapsone (b) Lidocaine (c) Hetrazan (d) Phenylbutazone

5. ----- is used in the treatment of dental surgery.

(a) Sulphamethazine (b) Sulphafurazole (c) Lidocaine (d) Chloroquine

6. Cobalt in vitamin B_{12} is found in which oxidation state?

(a) +1 (b) +2 (c) +3 (d) +5

7. Which metal is responsible for structural stability of DNA and RNA?

(a) Mg (b) Ca (c) Fe (d) Zn

8. Which protein transfers iron from stomach to various parts of the body ?

(a) Transferrin (b) Ferrochrome (c) Ferrioxamines (d) Given all

9. Valuable by-product obtain in ----- process during the production of sodium hydroxide.

(a) Causticizing (b) Lowing's (c) Electrolyte (d) None of them

10. ----- chemical is called "king of chemicals"

(a) $(NH_4)_2SO_4$ (b) NaOH (c) H_2SO_4 (d) HNO_3

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Q=2. Fill in the blanks:

1. The scale of δppm in CMR is ------ (0-50 / 0-beyond 200)

2. TMS is ------ (Volatile / Non -volatile)

3. Nirvanol drug is used as ------ (Anticonvulsant / Antimalerial)

4. Hetrazan drug is used as ----- (Anthelmintics / Antiarrhythmic)

5. ----- is important for bacterial nitrogen fixation (Zn / Mo)

- 6. Hemoglobin consists of ----- subunits.each containing one heme group (Four / Two)
- 7. Sodium hydroxide used as ------ (Mercerizing / Purification of gold)

8. Asbetos is used in ----- (Nelson cell / Castner- Kellner cell)

Q=3. Answer the following (Any ten).

- 1. Why TMS is used as a reference standard in NMR spectroscopy?
- 2. What is chemical shift?

.3. Discuss the shielded proton.

4. What are the requirement of an ideal drugs?

5.Discus the mode of action of antipyretic drugs.

6. Write the synthesis of drug which is used as a antianginal.

- 7. Draw the structure of A.T.P.
- 8. What is macronutrients?
- •9. Give the structure of Porphine.
- -10. Write the uses of HNO_{3.}
- 11. Discuss: Importance of H₂SO₄ in industries.
- 12. What is causticizing process.

Q=4. Answer the following (Any four)

.1.Discuss : diastereotopic proton and enantiotopic proton.

-2. Discuss proton-coupled and proton-decoupled spectrum with suitable example.

3. Write the synthesis and uses of the following drugs.

(a) Sulphafurazole (b) Sulphametazine



(20)

(08)

(32)

- 4. Define "Drugs". And discuss the mechanism of sulpha drugs.
- , 5. Write notes on vitamin B_{12} .
- 6. Write notes on Hemerythrin.
- 7. Write a chemical properties of H₂SO₄.
- 8. List different methods used for manufacture of nitric acid and discuss Birkland and Eyde process.

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