

Seat No.

[147] E+05 No. of printed pages: 4

Sardar Patel University B.Sc. EXAMINATION Fifth Semester (CBCS)

US05CCHE24 Analytical Chemistry

DATE: 26-11-2021

Time: 3:00 pm 5:00 Maximum marks 70

Multiple choice questions(10 marks)

LIBRARY

1)	Which type of metal is used in hydrogen electrode?
	a)Reactive b) Electropositive c) Inert d) Electronegative
2)	pH is expressed as
	a). $- \ln [H +]$ b) $- \log [H+]$ c) $+ \log [OH-]$ d) $- \log 10 [OH-]$
3)	In chromatography the efficiency of column increases with
	a) Height of column b) increase in column withd c)decrease in column width d) a & C
4)	To withstand high pressure , the column in HPLC is made up of
	a)I thick glass b) stainless steel c) both a & b d) None of them
5)	The percentage of halogenated pesticide in environment can be detected by
	a) TCD b) EC D c) FID d) FPP
6)	In gas chromatography quantitative determination is based on
	a) Number of peak b) Position of Peak
	C) Area of Peak d) All the above
7)	Electro Chemical detector belongs to detector
- \	a) Bulk property b) universal c) solute property d) Selective
8)	Continuous Extraction is used when the distribution ratio is very
01	a) high b) Moderate c) Low d) All of this
9)	For the study of distribution law the two solvent should be
401	a) miscible b) non-misible c) volatile d) Reactive
10)	Multiple Extraction isefficent than Single extraction
	a) more b) less c) equally d) none of the above

Fill in the blanks (8)

- 1) The pH range for glass electrode is.... (2 to 6/2 to 14)
- 2) To remove effect, platinised platinum electrode is used in conductame measu rements (polarization/dilution)
- 3) Ohms law can be representted as.... (I = E/R, E=I/R)
- 4) Number of theoretical plate refers to (height of column /measure of column efficiency)
- 5) The property of helium gas which is useful as a carrier gas is...(inertness/reactivity)
- 6) In gas chromatography stationary phase can be..........(liquid or gas /solid or liquid,)
- 7) Common examples of masking agent are (cyanides /carbonates)
- 8) Name apparatus used for the process of solvent extraction.....(separatory funnel/ bunchher funnel)

Short Que (Any 10) 10x2=20)

- 1) What are the advantages of the glass electrode ?
- 2) Give limitation of Quin hydrone electrode,
- 3) Define specific conductance,
- 4) Define Rf value for Paper Chromatography,

(P.T.O.)

- 5) Describe Soap bubble meter.
- 6) Define the term eluate and eluation in Column chromatography.
- 7) What are the funtion of solid support in GLC 1
- 8) What is the use of gaurd column in HPLC. 9
- 9) Enlist the specification for carrier gas in gas chromatography.
- 10) Give important applications of HPLC.
- 11) what is masking agent?
- 12) What are the limitation of Nerst Distribution Law

Long answer questions attempt any four (4x8=32)

- 1) Give advantage and disadvantages of hydrogen electrode.
- 2) Disuss the various types of paper chromatography in detail.
- 3) Discuss graphical method for determination of end point in acid-base.
- 4) Discuss the types of various ion Exchange Resin and give their applications,
- 5) Explain the principal and working of Thermal conductivity detector in Gas chromatography
- 6) Decribe the principle of HPLC. Describe the various components of HPLC.
- 7) Explain the process of solvent extraction. If ferentiate between distribution ratio and distribution coefficient.
- 8) Discuss in detail the factors which Influence the process of solvent extraction.

