Seat No.:



No. of Printed Pages: 02

SARDAR PATEL UNIVERSITY B. Sc. (SEMESTER-V) EXAMINATION Subject: Inorganic Chemistry (US05CCHE22)

Date: 24-11-2021 Day: Wednesday

[130/A-20] Eng.

> Time: 03:00 P.M. To 05:00 P.M. Total Marks: 70

Q:1 Answer the follow	wing multiple-choice q	uestions.		[10]
1. Which is the prin	cipal axis of rotation pr	esent in benzene mole	cule ?	Sei
$(a) C_6$ 2 Point group of P	$(b) C_4$	(c) C_3	(d) C_2 (d) C_2	. Durence
$(a) C_{a}$	(h)D			. 18
3 How many two-f	$(U)D_{2h}$	(c) D_{3h}	$(d) C_{3h}$	RARY
(a) 1	(1) 2	ent in cyclobutene mo	lecule ?	10/
(a) I	(0) 2	(c) 3	(d) 4	
4. CN IS a	_ neid ligand.	to Auronis eta Nationa		Nage
(a) strong	(b) weak	(c) bidentate	(d) octahedral	
5. Which orbital is v	ery important in CFT?			
(a) <i>s</i>	(b) <i>p</i>	(c) <i>d</i>	(d)f	
6. The complexes in	which the ligand substi	tution is fast are called	complexe	S.
(a) inert	(b) parallel	(c) octahedral	(d) labile	
7. S_N1 is known as	mechanis	m.		
(a) dissociation	(b) association	(c) both (a) & (b)	(d) none of these	
8. Greater stability of	chelated complexes is call	led		
(a) lability	(b) stability	(c) chelate effect	(d) none of these	
9. The molecular for	mula of inorganic rubbe	er is	(a) none of mese	
(a) [NPC1]n	(b) N3P3(NH2)6	(c) $(N_4P_4Cl_8)$	(d) [NPCb]nPCls	
10. How many isome	rs of S ₆ (NH) ₂ is possible	e?	(a) [1 (1 C12])II C13	
(a) 2	(b) 3	(c) 4	(d) 5	
			(u) 5	
Q:2 Fill in the blanks selecting the appropriate option given in the bracket:				[08]
1. The plane of refl	ection perpendicular to	the principal axis is ca	lled plane	
(horizontal / vert	ical)	1	province.	
2 molecule	has an infinite fold axi	s of symmetry. (Tetra	ahedral / Linear)	
3. In octahedral ligar	nd field <i>d</i> -orbital	possesses less energy	$(t_{2\alpha}/e_{\alpha})$	
4. The number of un	paired electron present	in $[Co(NH_3)_6]^{3+}$ is	(0 ne / zero)	
5. classified t	he complexes into labile	and inert complexes	(Toubo / Charles)	
6. The ability of a co	mplex to replace its one	e and mert complexes.	(Taube / Charles)	
(stability / lability)	of more figands is cal	iled its .	
7 Imides of sulphur	on he remains that here	1.0		
(S(NH)) = /S = 0	stury)	eneral formula	· · · · · · · · · · · · · · · · · · ·	
(Sn(1911)8-n / Sn-8(1	N(I)n)			
o IS no	of monomer of silicones	5. ($RS1(OH)_3 / C_6H_5S1(OH)_3$	Cl_3)	
Q:3 Short Answer Q	uestions (Attempt An	y Ten):		[20]
1. Construct the mul	tiplication table for C _{3v}	point group.		
2. Draw the diagram	to show that : $S_2 = i$	L		
3. Define: (a) Impro	per rotation (b) Identity	operation.		
4. Write note on spe	ctrochemical series.	10 1		

- 6. How the geometry of complex ion affected in the magnitude of Δ_0 ?
- 7. Give the limitations of Job's method.
- 8. Explain macrocyclic effect.
- 9. Mention factors affecting the stability of complexes depends on nature of ligand.
- 10. Give the uses of silicones.
- 11. Give the general properties of inorganic polymers.
- 12. What is nitride of sulphur?

Q:4 Long Answer Questions (Attempt Any Four):

- 1. Discuss D_{nh} and D_{nd} point group with proper example.
- 2. Give the symmetry elements and point group present in following:
 (a) Methane
 (b) SF₆
 (c) HCl
 (d) SO₂
- 3. Discuss the distribution of d^{x} electrons in high spin and low spin octahedral complexes.
- 4. Explain [CoF₆]³⁻ ion giving M.O. energy level diagram.
- 5. Discuss the factors affecting the stability of complexes depends on properties of central metal ion.
- 6. What is Acid Hydrolysis? Explain the mechanism of acid hydrolysis of octahedral complexes in which the inert ligand is a π acceptor.
- 7. Give the preparation, properties and structure of Borazine.
- 8. Give the preparation, properties and structure of Tetrasulphur tetranitride, S₄N₄.



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